

Active CO₂ Reservoir Management

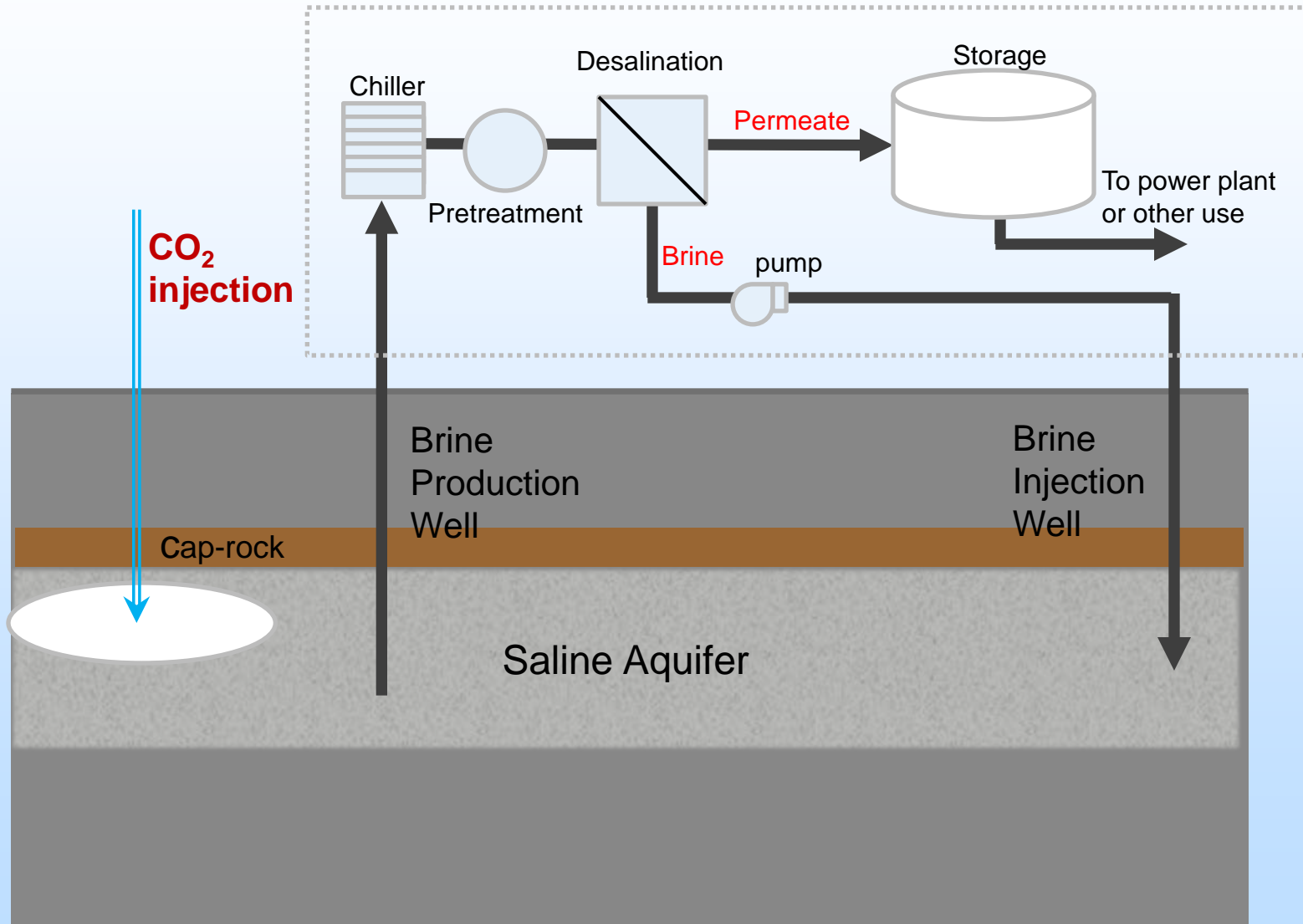
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U.S. Department of Energy
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Developing the Technologies and
Infrastructure for CCS
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Concept is to extract and desalinate aquifer brines to create fresh water and space for CO₂ storage



Presentation Outline

- Overview, Purpose, Goals and Benefits
- Technical status
 - Brine treatment and disposition
 - Reservoir management
- Accomplishments
- Summary and Planned work

Goals and Objectives

Technical Goals

Potential advantages of brine extraction:

- Allow reduction and active management of pressure in the subsurface
- Reduce the risks of cap rock failure and induced seismicity
- Provide a source of water for power plant cooling or other uses

Project Goals

- Use modeling to provide brine extraction/injection strategies that maximize CO₂ storage and minimize storage risk
- Identify technologies and cost estimates for brine disposition
- Provide quantitative input for overall cost-benefit analysis of brine extraction as a process used in carbon storage

Benefit to the Program

This project addresses all four program goals:

- Predict capacity
 - Brine removal affects/enhances storage capacity
- Assure permanent storage
 - Enhances cap-rock integrity and reduces induced seismicity
- Improve storage efficiency
 - Allows manipulation of sub-surface pressure field to maximize storage efficiency
- Best practices, especially site selection
 - Identifies preferred sites in terms of brine compositions

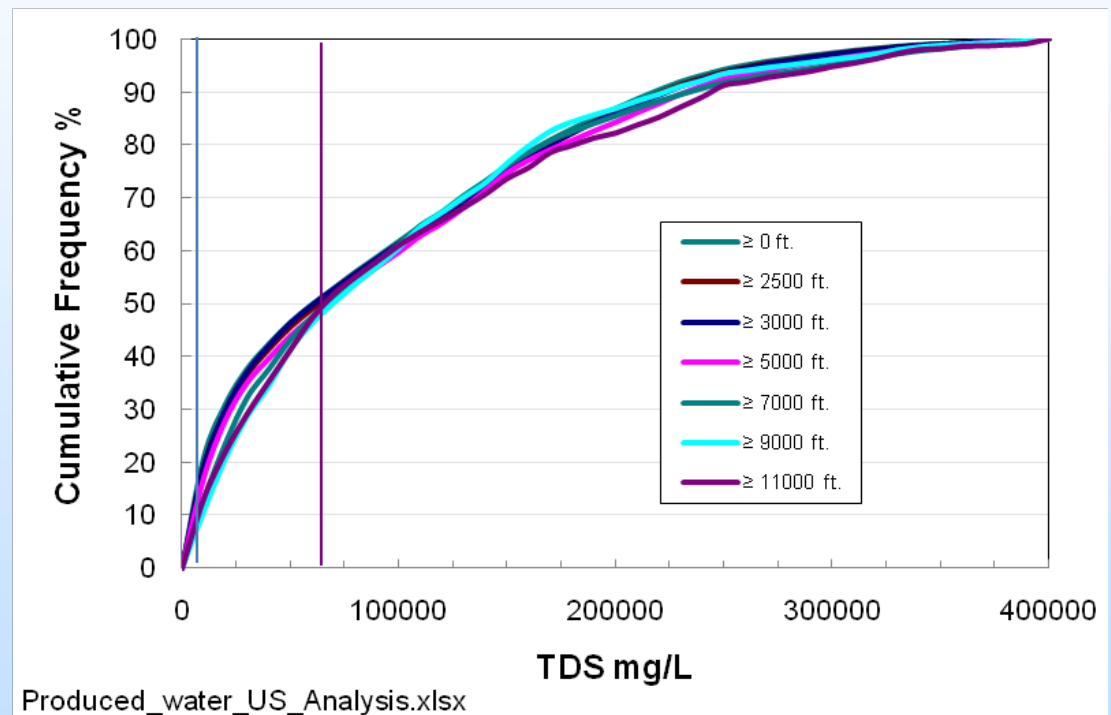
This project provides an analysis of brine extraction as a method for increasing the storage capacity and reducing the risk of failure at carbon storage sites.

Technical Status Outline

- Brine disposition
 - Expected brine compositions and characteristics
 - Appropriate desalination technologies
 - Analysis of membrane technologies
 - Markets for fresh/saline water derived from the brine
 - Estimated costs for desalination
- Reservoir engineering
 - Summary of progress on brine extraction/injection strategies

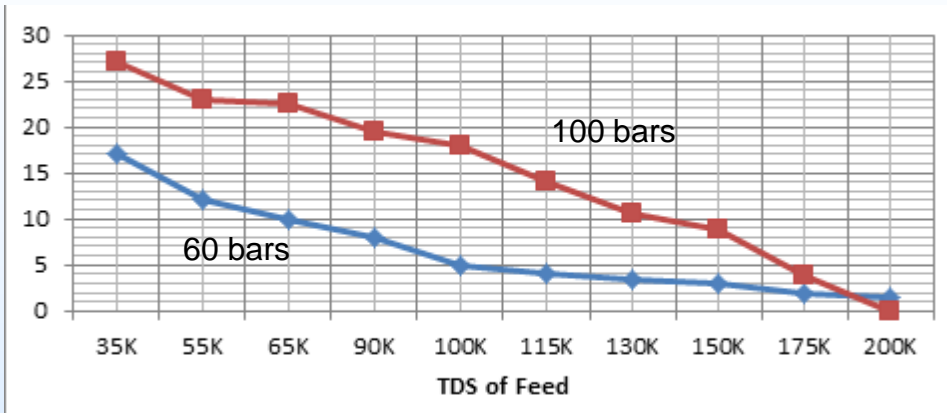
We should consider the compositions of formation fluids when choosing potential sequestration sites

- Formation fluids DO NOT become more saline with depth – to lower desal costs choose a site with lower salinity
- Membrane-based technologies are least expensive
- Thermal methods are needed to treat the highest salinity fluids (>20 wt% salt).
- High sulfate contents help enable nanofiltration to remove hardness in staged treatment
- Reservoir/brine temperatures up to at least 120°C are favorable for membrane desalination



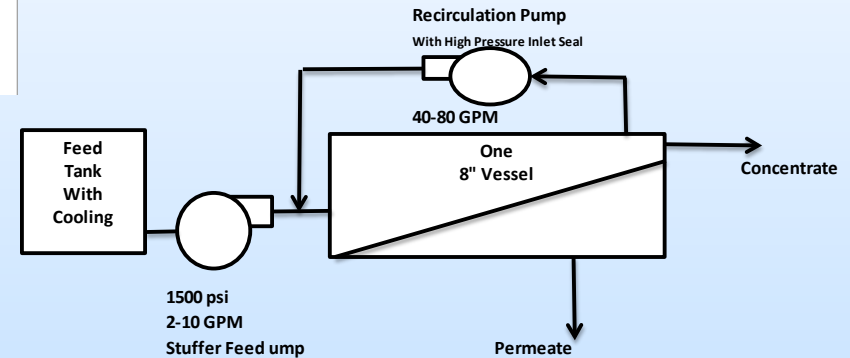
Bench-top tests of high pressure RO and NF were carried out and results used for cost estimation

Permeate flux (GFD) vs. Salinity



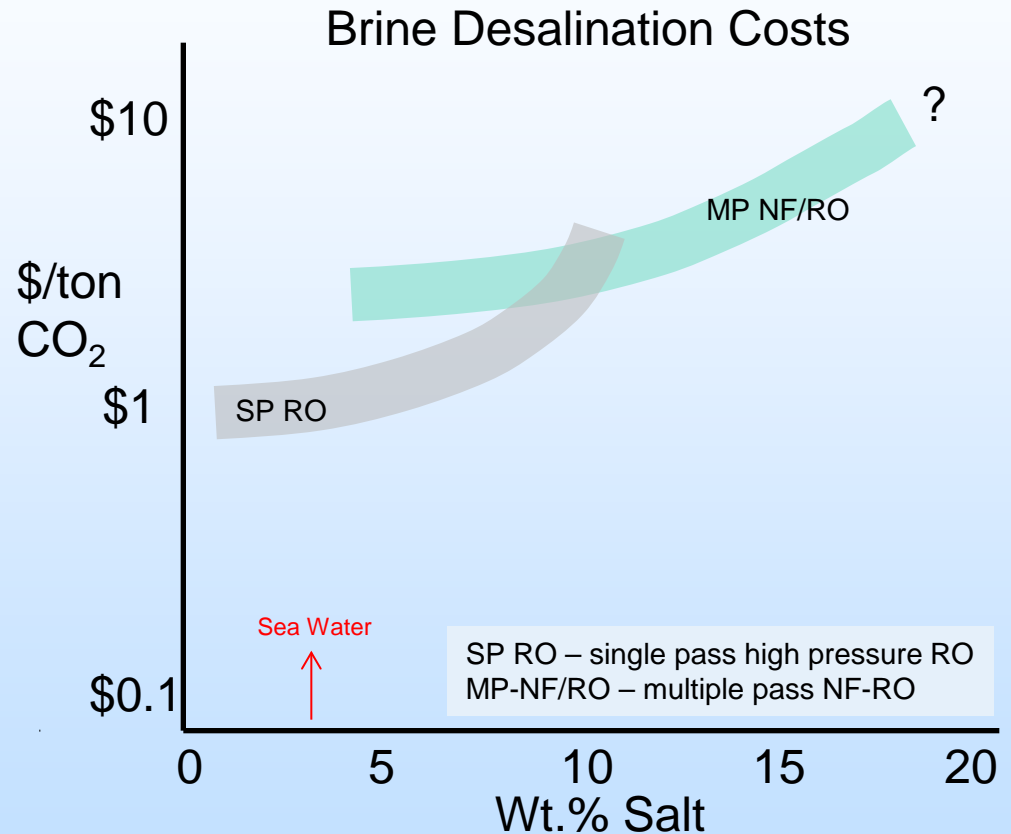
- Osmotic pressures increase with salinity
 - Sea water ~ 25-30 bars
 - 10 wt% brine ~ 80 bars
- Commercial membranes become impermeable at these pressures
- Staged treatment is possible using “loose” membranes that allow some salt passage

- Few data exist for desalination of brines more saline than sea water
- We carried out membrane desalination tests of brines up to 20 wt % to help extrapolate costs
- Membrane Development Specialists (MDS) carried out the tests



Staged treatment can extend the salinity range to cover brines up to 18 wt% - but at additional expense

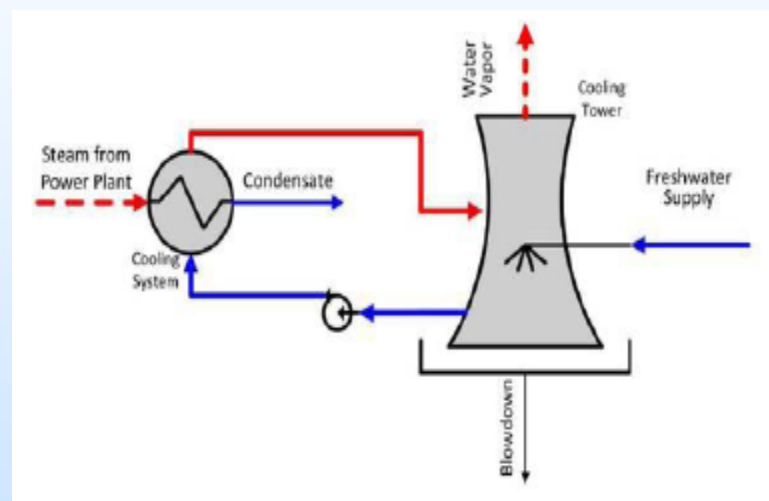
- Plot shows costs for surface treatment facilities
 - Does not include well-field costs
- Conversion to \$/ton assumes vol/vol of CO₂ at density 0.75cm³/g
- Single pass high pressure RO can desalinate brines up to about 8-10 wt%
- Multiple-pass NF-RO systems can extend this limit to almost 20 wt % but at substantial additional cost
- Costs are significant but not large compared to overall CCS costs



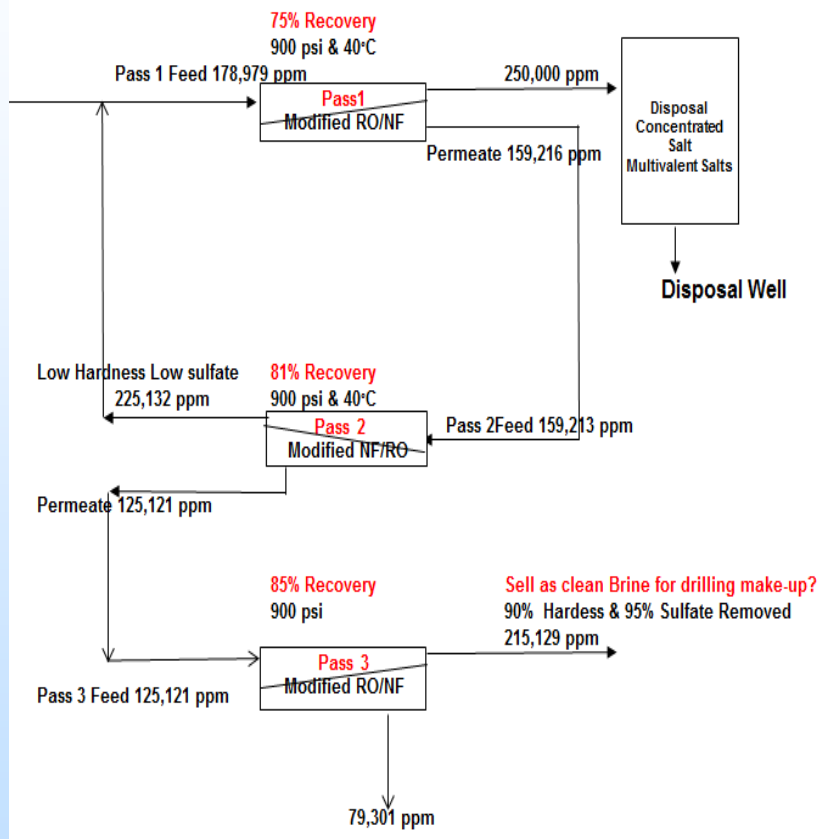
Rule of thumb: $\text{\$/ton CO}_2 = (\text{Brine Salinity} / \text{Sea water salinity})$

Multiple options are available for surface disposition of the treated saline water

- How much water?
 - 1 GW plant = 8000 acre-feet/y
 - Supply domestic needs of a town of 60,000 people
- That volume will supply about half the cooling water needed for a coal-powered plant
 - Important in arid regions and may make a power plant possible
- Other options:
 - Agricultural and potable water for local use
 - Saline cooling towers
 - Use softened water from NF
 - Saline oil-field make-up waters (11 wt % NaCl)
 - Use softened water from NF



Staged membrane treatment allows generation of tailored fluids for alternate uses



3-pass process flow diagram using NF membranes

- NF membranes separate monovalent from divalent species
- NF permeates are saline but lack hardness (Ca + Mg)
- Strategic combinations of RO and NF can generate useful saline fluids as well as potable water

Limits of membrane desalination technologies

- 10,000-40,000 mg/L: Standard RO with $\geq 50\%$ recovery
- 40,000-85,000 mg/L: Standard RO with $\geq 10\%$ recovery; higher recovery possible using 1500 psi RO membranes and/or multi-stage incremental desalination including nanofiltration
- 85,000-300,000 mg/L: Multi-stage process (NF + RO) using process design that may differ significantly from seawater systems
- > 300,000 mg/L: Not likely to be treatable

Overview of Pressure Management

- Strategies for pressure management using:
 - brine consumption via beneficial use
 - brine redistribution within a stack of saline aquifers (separated by impermeable seal units)
- Goals:
 - reduced risk of cap-rock fracturing and induced seismicity
 - suppressed CO₂ migration and leakage, reduced AOR
 - hydraulic isolation from neighboring subsurface activities
- Constraints
 - avoid CO₂ and brine breakthrough in well-field
 - limit well-field costs
 - Dual use wells
 - Vertical displacement strategies

CO₂ injection with horizontal wells - baseline

■ 30,000 km² semi-closed reservoir

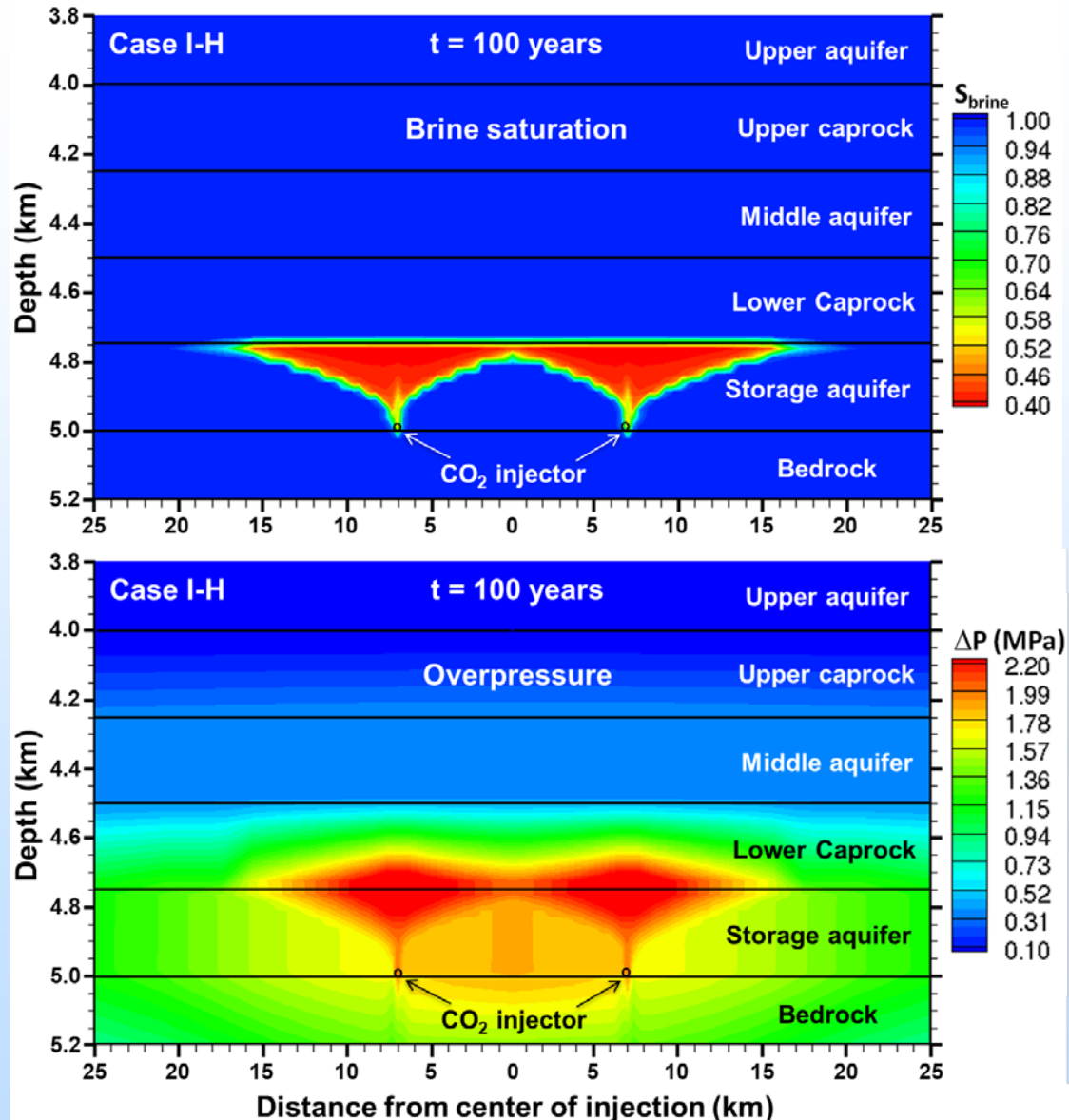
- similar in size to Illinois basin
- stack of 250-m-thick, 100-md saline aquifers and 10⁻³-md seal units
- bottom of storage aquifer at 2.5 km depth

■ 2 horizontal injectors

- running normal to the figures
- 4-km perforated length
- spaced 14 km apart at bottom of storage aquifer

■ CO₂ injection = 15 MT/yr

- CO₂ from 2 GWe of coal power plants
- 100 years of injection



Brine production and reinjection in storage aquifer

4 horizontal brine producers

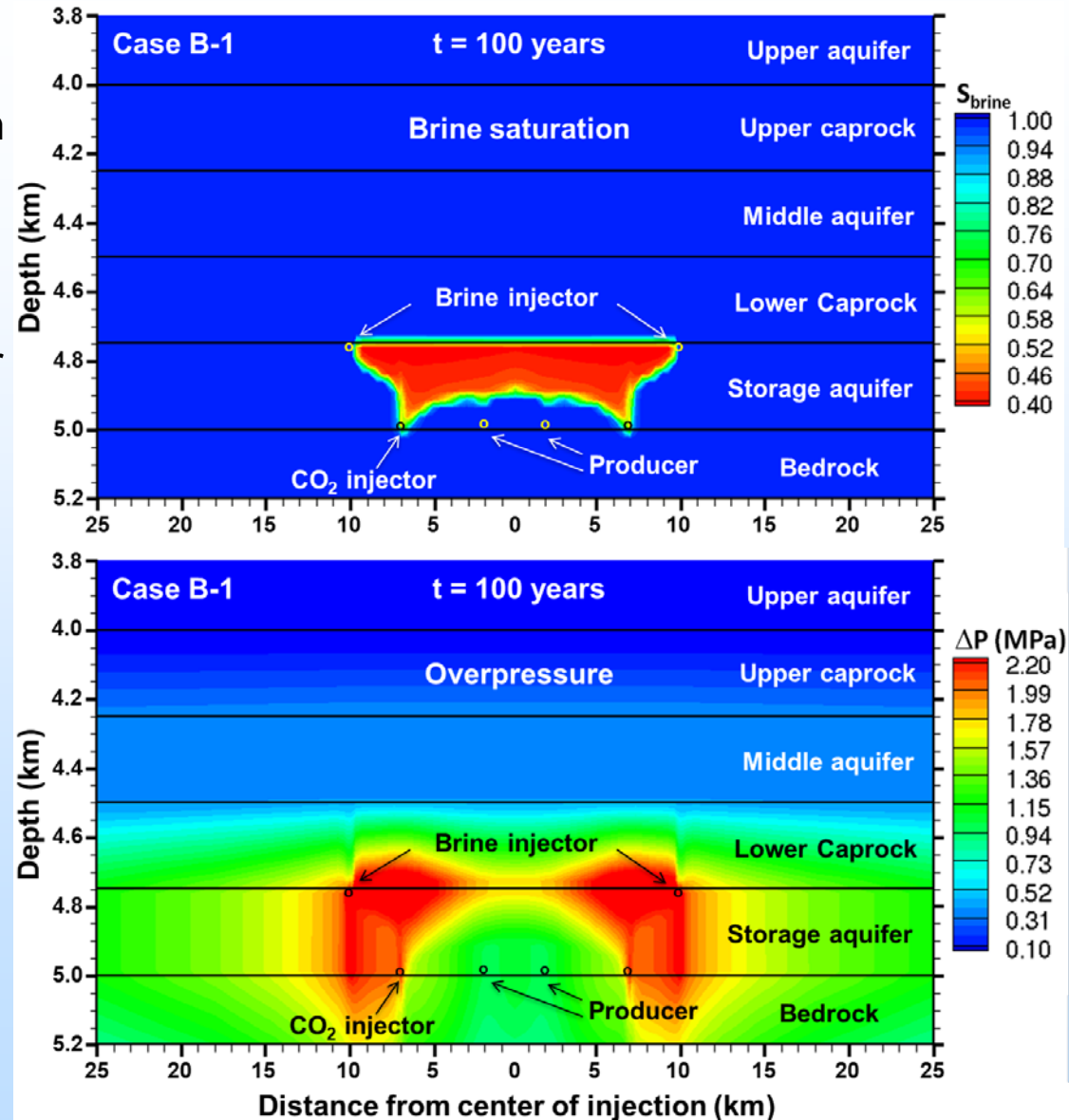
- at bottom of storage aquifer with 1-km perforated length
- spaced 4 km apart and 5 km from CO₂ injectors
- each producing 120 kg/sec for 100 yr

2 horizontal brine injectors

- 4-km perforated length at top of storage aquifer
- spaced 3 km from CO₂ injectors
- no brine consumption (reinject all produced brine)

Overpressure redistribution

- hydraulic barrier confines lateral CO₂ migration
- no changes above lower caprock or laterally in the far field



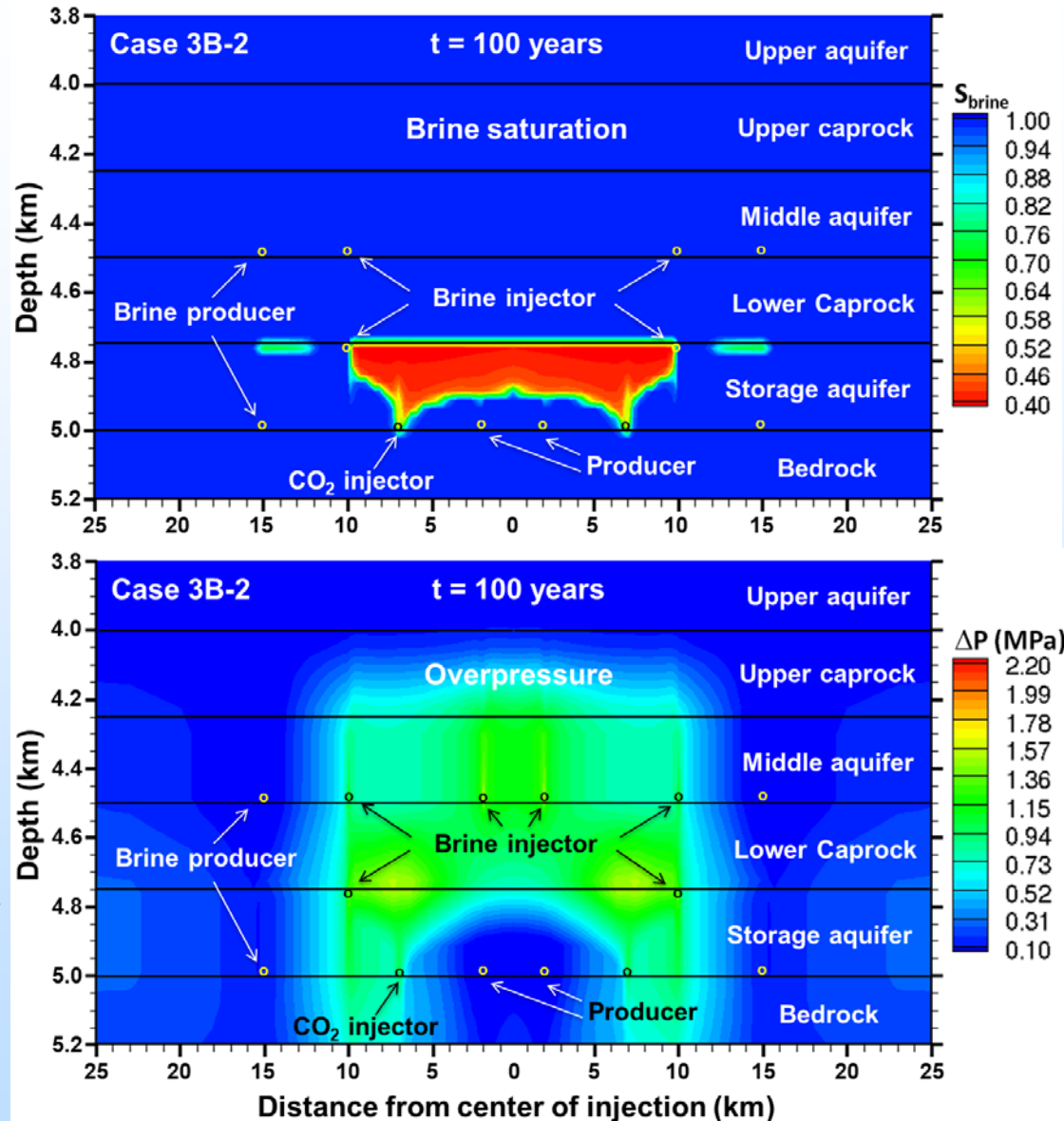
Brine production with desalination

16 horizontal brine producers

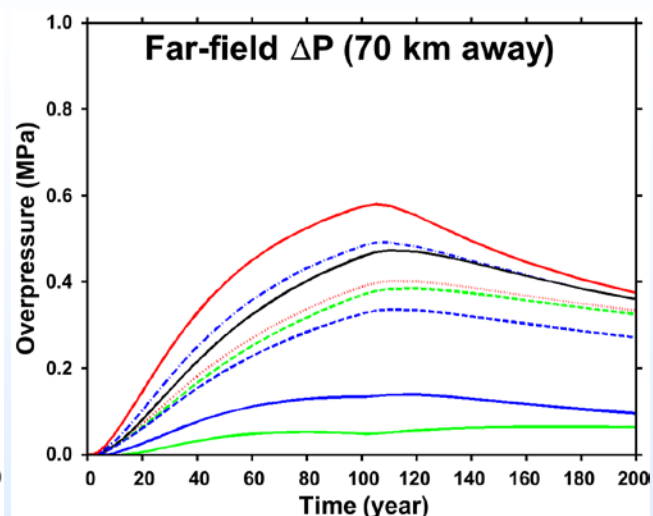
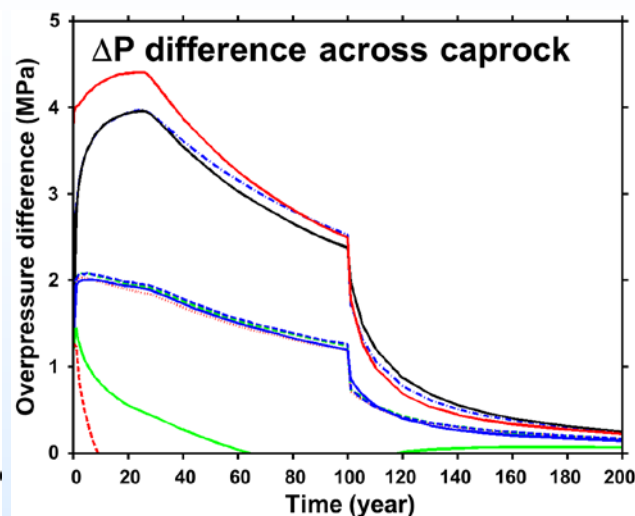
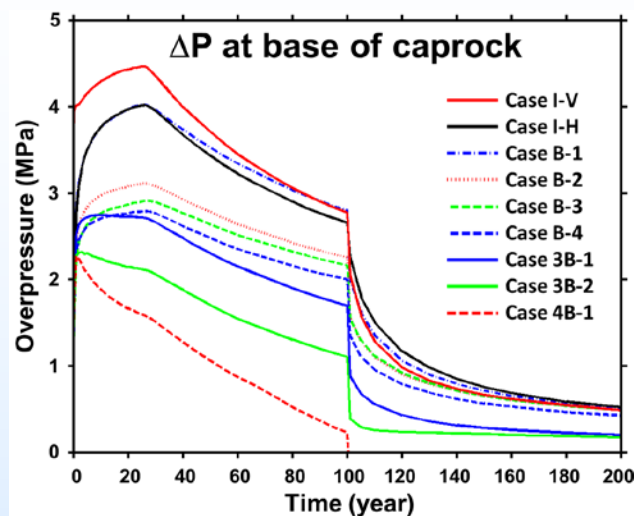
- 4 in middle of storage aquifer
- 8 at outer of storage aquifer
- 4 at outer of overlying aquifer
- 34% of produced brine (16,701 acre-ft/yr) is recovered for beneficial use
- 66% of produced (residual) brine is reinjected

Overpressure redistribution

- hydraulic barrier (ridge) suppresses CO₂ migration and leakage
- brine consumption creates a hydraulic trough, isolating the GCS operation from neighboring subsurface activities, reducing the Area of Review (AoR) and risk of induced seismicity



Overpressure reduction can be achieved by a combination of brine consumption and brine redistribution



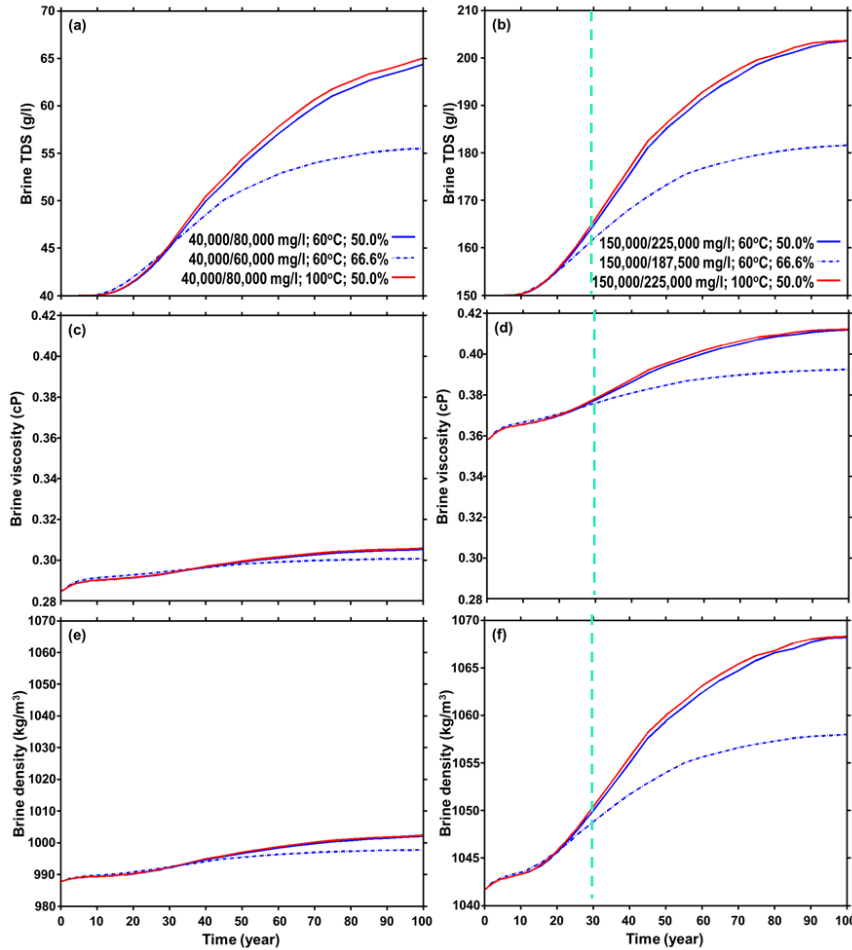
- A range of brine consumption and redistribution scenarios were considered
- Overpressure difference across caprock and far-field pressure perturbation can be nullified while reinjecting a majority (66%) of the produced brine

Case	Production/ injection mass ratio	Net mass (CO ₂ +H ₂ O) injection rate (kg/sec)	Product water generation rate (acre-ft/yr)	CO ₂ -injector-to-brine-injector spacing	Brine injectors in overlying saline aquifer	Brine-injection-to-production mass ratio	Number of brine producers	Notes
I-V	0	480	N/A	N/A	N/A	N/A	N/A	Vertical
I-H	0	480	N/A	N/A	N/A	N/A	N/A	Horizontal
B-1	1	480	0.0	3 km	0	1.0	0	Horizontal
B-2	1	480	0.0	3 km	2	1.0	0	Horizontal
B-3	1	480	0.0	6 km	2	1.0	0	Horizontal
B-4	1	316.8	4,175	3 km	2	0.66	0	Horizontal
3B-1	3	-9.6	12,526	3 km	4	0.66	8	Horizontal
3B-2	3	-9.6	12,526	3 km	4	0.66	8	Horizontal
4B-1	4	-172.8	16,701	3 km	4	0.66	12	Horizontal

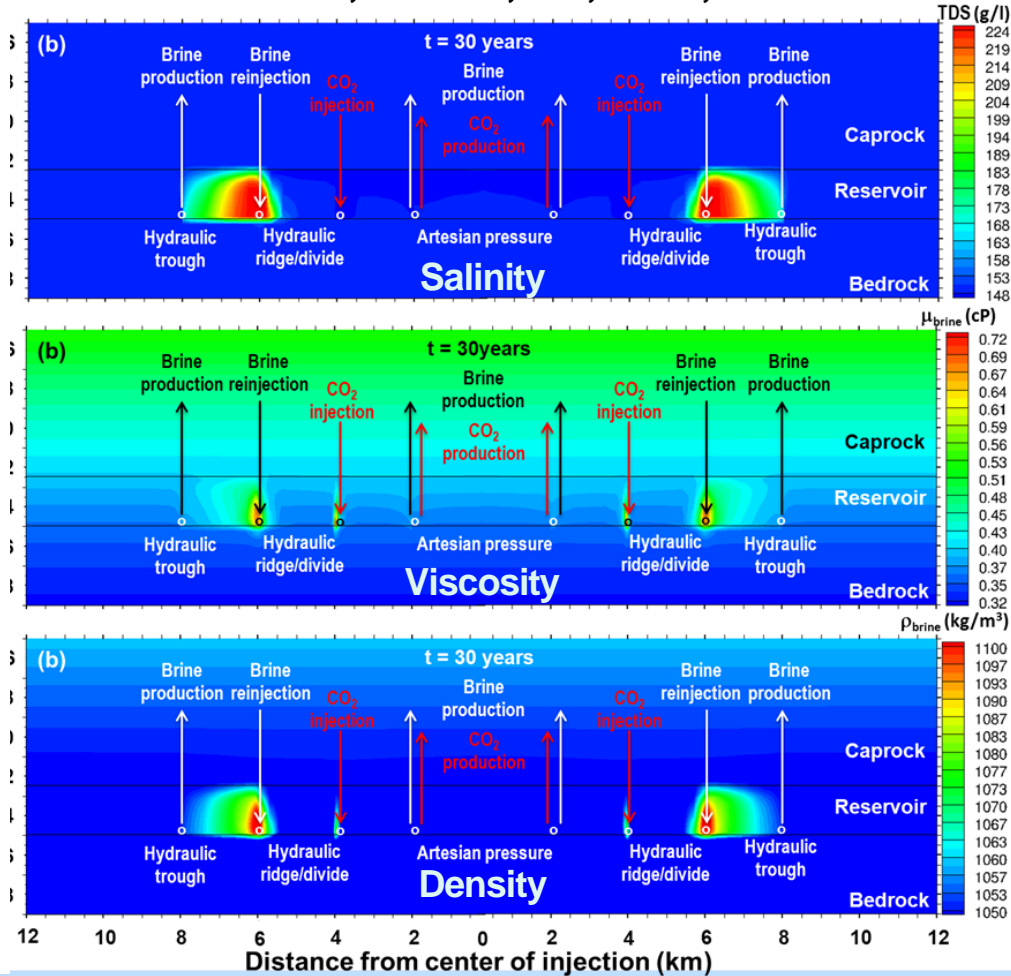
Residual-brine breakthrough occurs around 10 years for all cases considered

History at outer brine producers Cases 1-3

Cases 4-6



Vertical cross section at 30 years Case 4: 150,000/225,000; 60°C; 66.6%



Summary

Reservoir Management Task

- Pressure management can be achieved using a combination of
 - brine consumption
 - brine redistribution within a stack of saline aquifers separated by impermeable seals
- We have developed well patterns that achieve one or more of the following
 - reduced overpressure in the subsurface
 - creation of a hydraulic ridge to suppress CO₂ migration and leakage
 - creation of a hydraulic trough to isolate the GCS operation from neighboring subsurface activities and to limit pore-space competition and the AOR

Brine management task

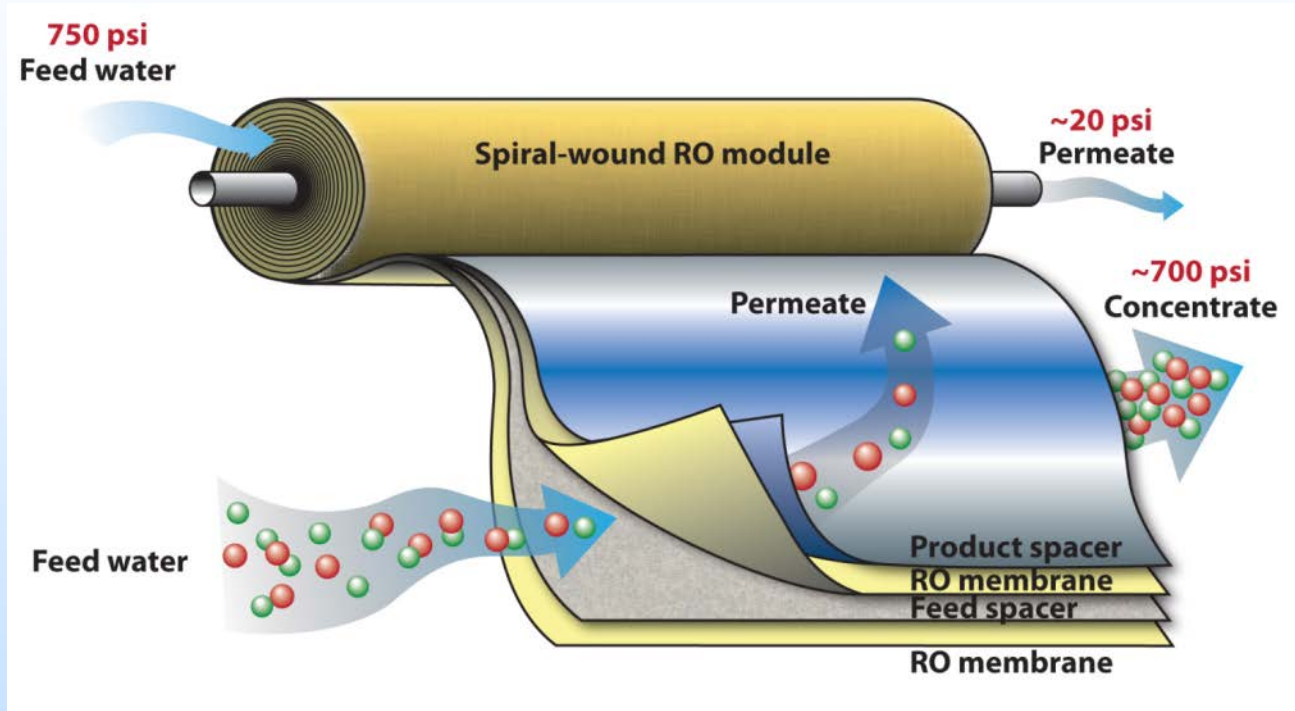
- We have completed an analysis of likely brine compositions
- Carried out laboratory work to enable desalination cost estimates for treatment of high salinity fluids

Future Plans

- Begin work to field test brine extraction
 - Select partner and site
- Re-focus reservoir modeling on simple systems
 - Minimize well-field costs while getting maximum pressure management benefit

A low-risk R&D effort could extend the range of RO technology to at least 150°C and 100 bars

The polyamide membrane is not the limiting factor

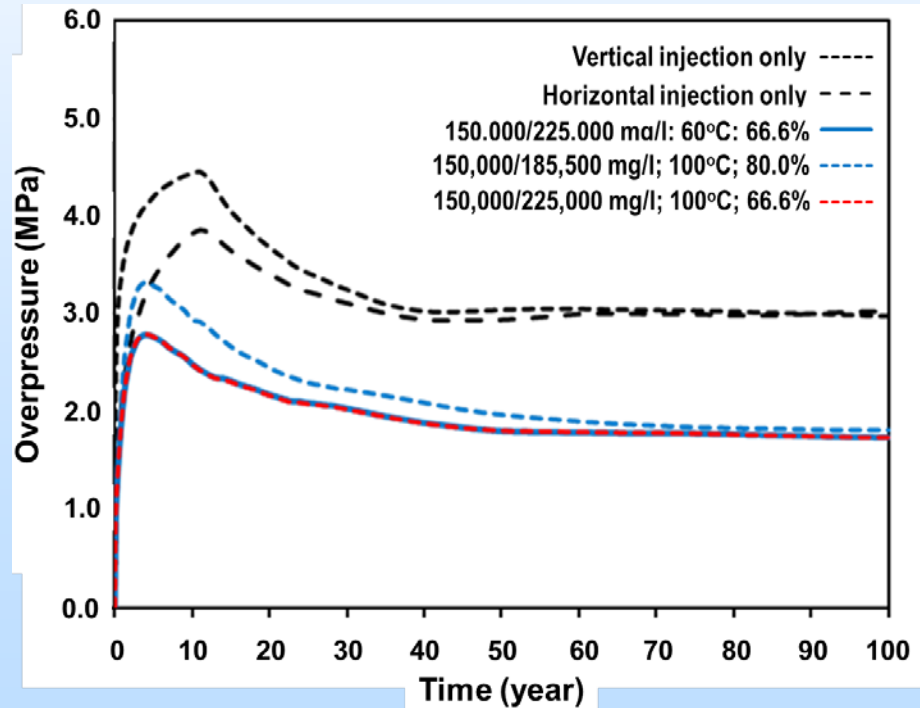


Allows desalination of brines with salinities up to 15 wt %

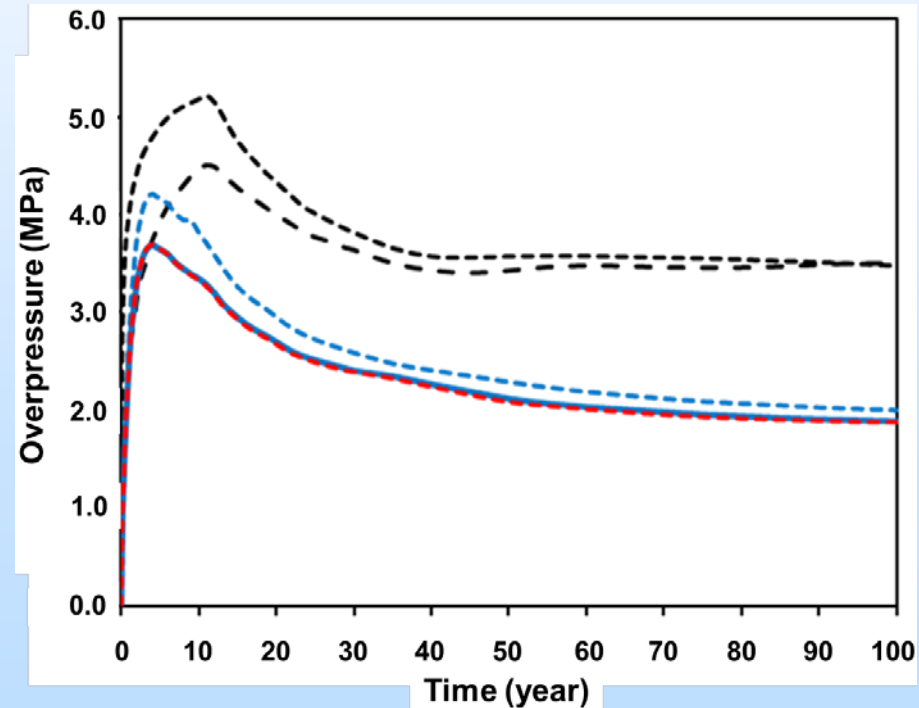
Brine redistribution can achieve substantial pressure reduction while reinjecting a majority of the residual brine

- Maximum overpressure is sensitive to formation salinity and temperature
 - increasing with salinity
 - decreasing with temperature
- Maximum overpressure is insensitive to salinity and temperature of injected residual brine
- Overpressure is reduced with increased brine consumption

Cases 1-3



Cases 4-6



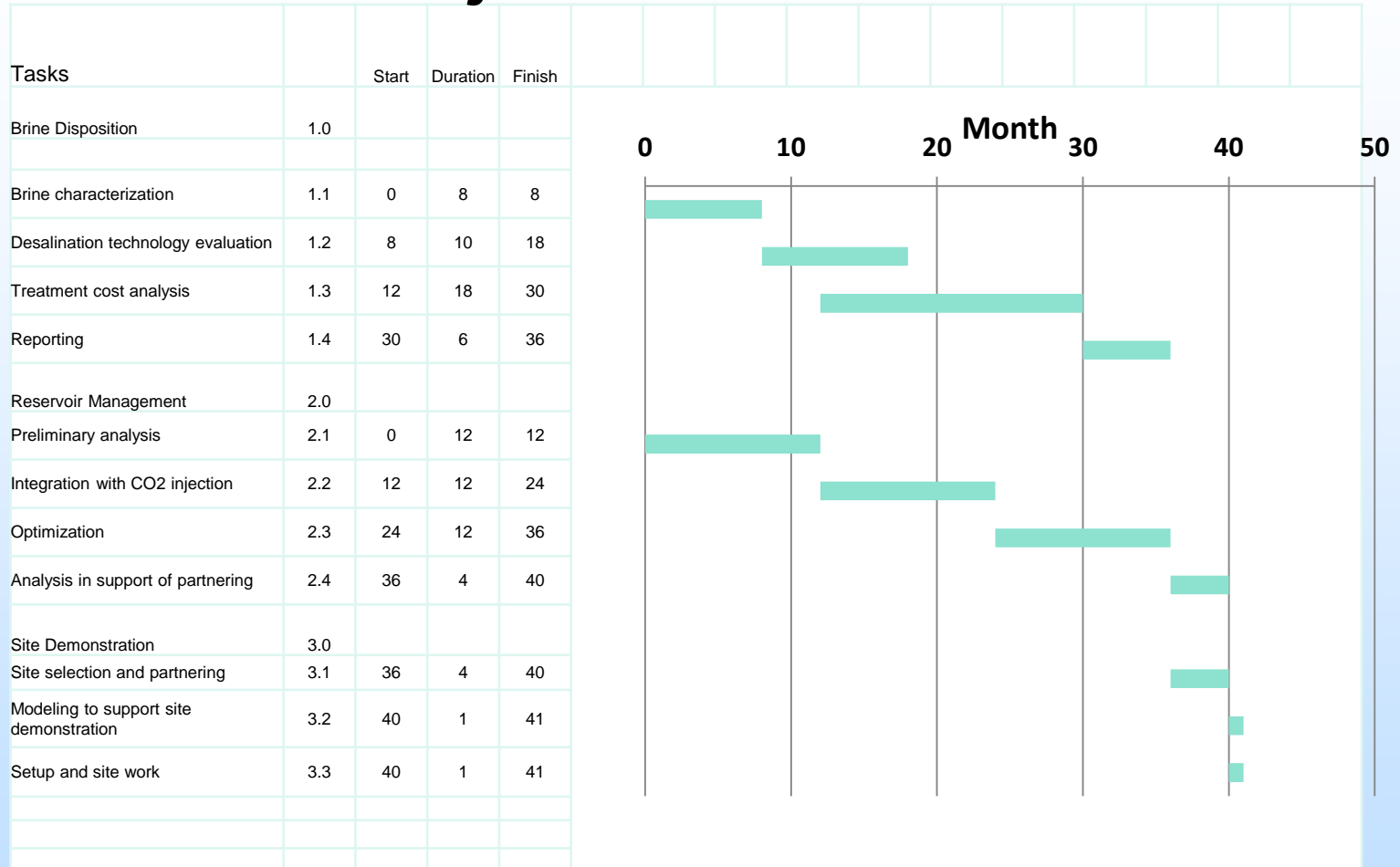
Accomplishments and Findings

- Finished analysis of expected brine compositions
 - Types and salinities
- Desalination technology selection
 - Membrane methods preferred over thermal
- Completed high pressure membrane desalination tests
 - Showed existing membrane technologies to be useful for fluids up to 175,000 ppm TDS
- Carried out modeling of simple to complex reservoir systems

Organization Chart

- Project team:
 - Lawrence Livermore National Laboratory
 - William Bourcier P.I. – brine disposition
 - Thomas Buscheck – reservoir modeling
 - Thomas Wolery (ret.) – brine disposition
 - Susan Carroll – Project Manager
 - Roger Aines – Carbon Program Leader
 - Membrane Development Specialists (MDS)
 - Subcontractor – Membrane desalination testing
 - Water System Specialists (in negotiation)
 - Subcontractor – Thermal desalination

Project Timeline



Bibliography

[Buscheck, T.A.](#), [Sun, YW.](#), [Hao](#) (2011) Combining Brine Extraction, Desalination, and Residual-Brine Reinjection with CO₂ Storage in Saline Formations: Implications for Pressure Management, Capacity, and Risk Mitigation

10th International Conference on Greenhouse Gas Control Technologies 01/2011; 4:4283-4290.

Buscheck, [T.A.](#), [Sun](#), [Hao](#), [Chen](#), [Court](#), [Celia](#), [Bourcier](#), [W.L.](#), [T.J. Wolery](#) (2011) [Geothermal energy production from actively-managed CO₂ storage in saline formations](#) Proceedings of the Geothermal Resources Council 35th Annual Meeting; 10/2011

[Thomas A Buscheck](#), [Yunwei Sun](#), [Mingjie Chen](#), [Yue Hao](#), [Thomas J Wolery](#), [S Julio Friedmann](#), [Roger D Aines](#), (2012) [Lawrence Livermore, National Laboratory](#) Active CO₂ Reservoir Management for CO₂ Capture, Utilization, and Storage: An Approach to Improve CO₂ Storage Capacity and to Reduce Risk, In Proceedings for Carbon Management Technology Conference.

[Roger D. Aines](#), [Thomas J. Wolery](#), [William L. Bourcier](#), [Thomas Wolfe](#), [Chris Hausmann](#) (2011) Fresh water generation from aquifer-pressured carbon storage: Feasibility of treating saline formation waters [Energy Procedia](#) 4:2269-2276. DOI:10.1016/j.egypro.2011.02.116.

Bourcier, W. L., Wolery, T. J., T. Wolfe, C. Hausmann, T. A. Buscheck, and R. D. Aines (2011) A preliminary cost and engineering estimate for desalinating produced formation water associated with carbon dioxide capture and storage. Int. J. Greenhouse Gas Control, v. 5. Sept 1, 2011.